Microbial biodegradation of microcystin-RR by bacterium *Sphingopyxis* sp. USTB-05

Mulan Zhang¹, Gang Pan^{1,*}, Hai Yan²

1. State Key Laboratory of Environmental Aquatic Chemistry, Research Center for Eco-Environmental Sciences,
Chinese Academy of Sciences,

Beijing 100085, China. E-mail: zhangmulanmine@yahoo.com.cn

2. Department of Biological Science and Technology, School of Applied Science,
University of Science and Technology Beijing, Beijing 100083, China

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Abstract

A strain, USTB-05, isolated from Lake Dianchi, China, degraded the cyanobacterial toxin microcystin-RR (MC-RR) at the rate of 16.7 mg/L per day. Analysis of 16S rDNA sequence showed that the strain was *Sphingopyxis* sp. Enzymatic degradation pathways for MC-RR by *Sphingopyxis* sp. USTB-05 were identified. Adda-Arg peptide bond of MC-RR was cleaved and then a hydrogen and a hydroxyl were combined onto the NH₂ group of Adda and the carboxyl group of arginine to form a linear molecule as intermediate product within the first few hours. Then, through dehydration reaction, two hydrogen of amino group on arginine reacted with adjacent hydroxyl on carbon to form a linear MC-RR with two small peptide rings as the final product after 24 hr. These biodegradation pathways were different from those reported for other strains, implying that MC-RR may undergo different transformations and different products were formed due to various bacteria in natural lakes and reservoirs.

Key words: microcystin-RR; biodegradation; *Sphingopyxis* sp. USTB-05; enzymatic pathways; Lake Dianchi

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Introduction

Toxic blooms of blue-green algae occur worldwide in eutrophic lakes, ponds and reservoirs (Paerl et al., 2001). Harmful algae, such as *Microcystis*, *Anabaena*, *Nostoc* and *Aphanizomenon*, can produce toxins, in which microcystins (MCs) are the most commonly detected toxins. MCs are a group of cyclic heptapeptide hepatotoxins, and show hepatotoxicity and tumor-promoting activity by inhibition of protein serine/threonine phosphatases 1 and 2A (Dawson, 1998). MCs are usually retained in algal cells during the growth. These intracellular toxins are eventually released into water bodies during senescence and decomposition processes, which could cause severe health risks to animals and human beings.

MCs are chemically stable in natural water (Jones and Orr, 1994; Tsuji et al., 1994). The concentrations of MCs in many natural lakes are much less than the estimated levels produced by cyanobacteria (Tsuji et al., 1996), such as in Lake Dianchi in Yunnan Province of China, which has been hypertrophicated in recent years, and harmful algae blooms appear all year around. In previous studies, we found that algae blooms in Lake Dianchi could produce high content of MCs (Yan et al., 2002, 2004a; Zhang

et al., 2004), where MC-RR has the highest occurrence. However, the dissolved MCs concentration in the water of Lake Dianchi was generally low (less than 1 µg/L), even when the concentration of algal cells exceeded 10⁹ cells/L (Zhang et al., 2003; Pan et al., 2006). Several pathways have been proposed to elucidate the reduction of MCs, in which photolysis and biodegradation play important roles. Photo-degradation is important in removing the dissolved MCs (Zhang et al., 2003; Song et al., 2007). However, ultraviolet can hardly penetrate even a thin layer of colored water, let alone water containing high dense algae. Biodegradation is therefore an important process controlling the fate of MCs in natural waters (Chen et al., 2008).

The bacterial communities occurred in the water body where harmful algae blooms are common, can degrade MCs along with other organic compounds which are frequently found in the environment. The isolation and biodegradability of many pure bacterial strains, such as *Sphingomonas* sp., *Sphingopyxis* sp., *Pseudomonas aeruginosa*, *Paucibacter toxinivorans*, *Delftia acidovorans* and *Burkholderia* sp., from Australian reservoirs, Argentinean reservoirs, Japanese lakes, Finland lakes, China lakes and Brazilian coastal lagoon, respectively have been reported (Jones et al., 1994; Takenaka and Watanabe, 1997; Ishii et al., 2004; Rapala et al., 2005; Tsuji et al., 2006; Amé

* Corresponding author. E-mail: gpan@rcees.ac.cn

et al., 2006; Ho et al., 2007; He et al., 2006; Lemes et al., 2008). Most of strains reported belong to genus *Sphingomonas*, such as 7CY, B-9, Y2, ACM-3962, CBA4 and LH21 (Jones et al., 1994; Ishii et al., 2004; Amé et al., 2006; Ho et al., 2007; Tsuji et al., 2006). Among these strains, *Sphingomonas* sp. Y2 has the highest degradation rate (13.0 mg/L per day) for microcystin-RR (MC-RR) (Park et al., 2001).

To understand the biodegradation pathways for MC-RR, the previous studies focused on the biodegradation pathways of *Sphingomonas* sp. B-9 and *Sphingomonas* sp. CBA4. *Sphingomonas* sp. B-9 degraded MC-RR by cleaving the Adda-arginine peptide bond and the cyclized structure was turned into linear as intermediate product. Then the linear MC-RR was divided into tetrapeptide as another intermediate product. The tetrapeptide was finally degraded into Adda (Imanishi et al., 2005). The MC-RR biodegradation pathways by *Sphingomonas* sp. CBA4 would start by demethylating MC-RR, affording the intermediate product, which is finally biodegraded within 72 hrs (Amé et al., 2006). However, whether the biodegradability and pathways of MC-RR by other strains including *Sphingomonas* sp. are different, and how they contribute to the fate of algae toxins in natural freshwaters are still poorly understood.

In the present study, we isolated a strain *Sphingopyxis* sp. USTB-05 from Lake Dianchi, which showed high ability in degrading MC-RR. Its enzymatic degradation pathways on MC-RR were determined. Both intermediate and final degradation products were identified using a combination of HPLC, LC-MS and MALDI-TOF MS.

1 Materials and methods

1.1 Isolation of bacterium

The strain for the biodegradation of MC-RR was isolated from the sediment samples collected from Lake Dianchi. Five gram of sediment samples were re-suspended in sterilized water for 2 hr. The supernatant was used as inoculums and subjected to medium containing (1 L distilled water): 1.0 g $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 0.5 g KH_2PO_4 , 4.0 g K_2HPO_4 , 1.0 g NaCl, 20 mg CaCl_2 , 5 mg FeSO_4 , 5 mg ZnCl_2 , 5 mg $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$, and 0.5 mg CuCl_2 . The medium was sterilized before use and the pH was adjusted to 7.0. MC-RR was added as the sole carbon and nitrogen source at the initial concentration of 50 mg/L. The mixture was incubated for 3 days at 30°C in 50 mL sterile flasks. After three subcultures were prepared by transferring 2.5% (V/V) of the starting inoculums to another flask containing medium with MC-RR, the incubated inoculums were spread on a solid medium, in which 1% agar (V/V) was added to medium. After several days, different types of colonies were isolated and transferred to medium containing MC-RR (50 mg/L) to determine their biodegradability. After 3 days of incubation, the solution was centrifuged, and the concentration of MC-RR in the supernatant was measured. Among 7 isolated strains, the one that had the highest biodegradability was incubated in

axenic medium.

1.2 Identification of strain

The isolated strain was identified using 16S rDNA analysis with Blast. 16S rDNA was selectively amplified with PCR (Eppendorf-5331) for a total of 30 cycles at 94°C for 1 min, then 50°C for 1 min and finally 72°C for 3 min. The upstream primer was 5'-AGAGTTTGATCATGGCTCAG-3' and the downstream primer was 5'-CTACGGTTACCTTGTTACGAC-3'. The amplified DNA was sent to the Beijing Sunbiotech Co. (China) for sequencing. A multiple alignment was obtained and a phylogenetic tree was constructed according to the neighbor-joining method by Clustal software (Saitou and Nei, 1987).

1.3 MC-RR

Standard MC-RR (95% purity, molecular weight (MW) 1038.2) was purchased from Sigma Chemical Co., USA, and used to test the biodegradation of MC-RR by enzymes of the strain and used as a standard to establish the calibration curve. For the biodegradation kinetic experiment where large amount of MC-RR was needed, the extracted MC-RR with the purity of 7.0% from cyanobacterial cells collected from Dianchi Lake was used. The content of MC-RR in dry cyanobacterial cells was 1.3 mg/g. The extraction method of MC-RR was described elsewhere (Zhang et al., 2004; Yan et al., 2004b).

1.4 Preparation of strain and its intracellular enzyme

After 3 days of incubation, cells of the strain were harvested by centrifugation at 12,000 r/min for 10 min and washed three times with 50 mmol/L potassium phosphate buffer (pH 7.0). Then the cells were resuspended using potassium phosphate buffer to make a stock solution containing 0.81 mg cells/mL (dry weight) and stored at -18°C. This strain solution was used for the biodegradation kinetic experiment.

To obtain the intracellular enzyme, 26 mg cells (dry weight) harvested were resuspended in 20 mL potassium phosphate buffer solution and disrupted using an ultrasonic disruptor with an output power of 600 W for 15 min at 4°C. The cell debris was removed by centrifugation at 18,000 r/min for 20 min, and the supernatant was used as the intracellular enzyme in MC-RR biodegradation pathways studies and stored at -18°C.

1.5 Biodegradation kinetic

Cells of the strain (0.0015 mg) were added to a 50-mL flask containing 15 mL potassium phosphate buffer (50 mmol/L) in the presence of 50.2 mg/L MC-RR. The flask was shaken at 150 r/min and at 30°C. At the reaction time of 0, 1, 2 and 3 days, aliquot samples were taken and centrifuged at 12,000 r/min for 10 min. Concentration of MC-RR in the supernatant was measured using high performance liquid chromatography (HPLC, Shimadzu LC-10A, Shimadzu Co., Japan). The experiments were conducted in duplicates and the average values were used for analysis. A bacterial-free medium was used as a control.

1.6 Enzymatic degradation pathway

In a 5 mL centrifuge tube, 15 mg/L MC-RR and 38.4 mg/L (protein content) intracellular enzyme were prepared in potassium phosphate buffer. The total reaction volume was 1 mL. The tube was shaken at 150 r/min and at 30°C. Samples (150 μ L) were taken from the reaction vessel at 0, 20, 60, 120, 300 and 1440 min. Hydrochloric acid (1%, V/V) was added to the samples to stop the biodegradation reactions. The samples were measured using HPLC.

In order to identify the intermediate products, 0.5 mL of samples were taken at 20 and 300 min from another duplicated reaction vessel, which was concentrated using C18 solid-phase extraction cartridge (Waters, OASISTMHLB, USA, 30 mg/mL). Then methanol was used to elute the concentrated products with the rate of 1 mL/min. The elution was analyzed using liquid chromatography tandem one mass spectrometry (LC-MS, API-3000, Applied Biosystems, USA) to determine the molecular weight (MW) of parent MC-RR and its degradation products.

To further confirm the MW of the final product that cannot be identified using LC-MS in the previous procedures, a solution of mobile phase (0.3 mL) was collected immediately after target peaks appeared in UV chromatogram. The collected solution was dried with nitrogen to remove the acetonitrile contained in the mobile phase. Solids obtained (containing MC-RR and its degradation products) were dissolved with distilled water, and analyzed using matrix-assisted laser desorption/ionization time-of-flight mass spectrometry (MALDI-TOF MS, BIFLEX III, Bruker Inc., USA).

1.7 Analysis methods

For MC-RR analysis, HPLC was operated at 239 nm. The mobile phase was 40% (volume percent) acetonitrile-water solution containing 0.03% (volume percent) of trifluoroacetic acid. The flow-rate was 1.0 mL/min and the injection amount was 20 μ L. A Waters Bondapak C18 column (300 $\times$ 3.9 mm) was used to separate MC-RR and its degradation products.

A 3-micron YMC ODS AQ analytical column from Waters Corp. (USA) was used in the chromatographic measurement in LC-MS analysis. For LC detection, the analysis method was the same as that of HPLC. For MS detection, precursor ions for samples and internal standards were determined from mass spectrum obtained during infusion into the API-3000 mass spectrometer. Using API turbo ionspray (TISP) source, the mass spectrometer was operated in the positive ionization mode with the collision gas off.

For MALDI-TOF MS analysis, the instrument was equipped with a nitrogen laser operating at 337 nm, and a two-state ion source operating in the delayed extraction mode. MALDI-TOF MS was operated in CCA positive ionization mode. Briefly, 2 μ L of a saturated solution of *a*-cyano-4-hydroxycinnamic acid (acetonitrile/0.1% trifluoroacetic acid, 3:2, V/V) was mixed with 2 μ L of enzymatic biodegradation samples. The preparation of the enzymatic biodegradation samples was described in the

above section. Afterward, 2 μ L of the mixture was applied to the sample plate, and air-dried at 24°C. Measurements were performed at an acceleration voltage of 20 kV using reflector mode, allowing the determination of monoisotopic mass values. Each spectrum refers to the sum of 100–200 individual laser shots.

2 Results and discussion

2.1 Strain identification

16S rDNA sequence homology searches using the basic local alignment search tool (BLAST) system corroborated the positioning of USTB-05 in genus *Sphingopyxis* sp. The neighbor-joining phylogenetic tree is shown in Fig. 1. The close relatives were proved to be *Sphingomonas* sp. and *Novosphingobium* sp. The homology percentage of 16S rRNA in the strain USTB-05 and *Sphingopyxis* sp. C-1 was 96%. Although it is reported that *Sphingomonas* sp. ACM3962, 7YC, CBA4 and Y2 are able to degrade MCs (Jones et al., 1994; Ishii et al., 2004; Amé et al., 2006; Tsuji et al., 2006), the position of the strain in the tree was distinctively separated from the clusters of them. This indicated that USTB-05 isolated in this study may be a new species of *Sphingopyxis* for MCs degradation. The complete sequences of the 1453 bp 16S rDNA fragment from this strain had been deposited in the GenBank database under accession number EF607053.

2.2 Microbial biodegradation kinetics

The kinetic curve of the biodegradation of MC-RR by *Sphingopyxis* sp. USTB-05 is shown in Fig. 2. After one day of lag phase, the concentration of MC-RR decreased rapidly, and the initial MC-RR of 50.2 mg/L was completely removed within 3 days, while the MC-RR in control flask remained relatively constant. The average biodegradation rate of MC-RR by the strain was 16.7 mg/L per day which showed that the strain had a strong ability in the degradation of MC-RR.

2.3 Enzymatic degradation pathway

Different degradation products of MC-RR at 0, 20, 60, 300, and 1440 min were clarified based on the peak position in Fig. 3. For the sample collected at 0 min, the peak at 5.3 min was MC-RR because it appeared at the retention time of standard MC-RR (Yan et al., 2002). The first peak at 3.0 min was due to an impurity of the MC-RR solution. For the sample collected at 20 min, the peak of MC-RR decreased with reaction time, and a new peak I at 3.8 min was observed. For the sample collected at 120 min, another peak II at 8.10 min was observed while the peak of MC-RR disappeared. After reached a maximum at 60 min, peak I decreased and disappeared after 24 hr, indicating that it was an intermediate product, whereas, peak II increased with time and kept constant after 24 hr, indicating that it was the final biodegradation product.

To identify the degradation products, 0.5 mL samples were taken from another biodegradation reaction vessel at reaction time of 20 and 300 min. Because MC-RR, peak

Fig. 1 Neighbor-joining phylogenetic tree showing the position of the strain based on 16S rDNA gene sequence. Database accession numbers of reference sequences are shown after the strain names. The value on each branch is the estimated confidence limit (expressed as a percentage) for the position of the branch as determined by a bootstrap analysis.

Fig. 2 Bacterial biodegradation kinetics of MC-RR by *Sphingopyxis* sp. USTB-05.

I and peak II were baseline separated with the retention times of 5.3, 3.8 and 8.0 min, respectively. Samples collected at different reaction times were mixed together to measure the intermediate and final degradation products contained in peak I and II by LC-MS. Results are presented in Fig. 4.

Four distinct peaks were observed in total ion chromatogram (TIC) (Fig. 4a). Based on the retention time in

UV chromatogram (Fig. 3 and Fig. 4b), the second and third peaks marked with column in TIC corresponded to MC-RR (retention time 4.7 min) and peak I (retention time 3.4 min), respectively. The MS spectrum of MC-RR in Fig. 4c revealed a major ion at m/z 1038.7, corresponding to the protonated molecular ion $[M+H]^+$. The MS spectrum for peak I (retention time 3.58 min) showed four ion peaks at m/z 1056.9, 529.1, 905.6, 453.5 (Fig. 4d). Due to the different degrees of protonation, ion peaks at m/z 1056.9 and 529.1 represented one product, corresponding to the protonated molecular ion $[M+18+H]^+$ and $[M+18+2H]^{2+}$, respectively. Ion peaks at m/z 905.6 and 453.5 represented another product, corresponding to the protonated molecular ion $[M-133+H]^+$ and $[M-133+2H]^{2+}$, respectively.

Since the final products contained in peak II can only be detected in UV chromatogram (Fig. 4b), but not clearly in TIC spectrum (Fig. 4a), sample of 24 hr was further analyzed using MALDI-TOF-MS. Results in Fig. 5 showed three ions with m/z of 991.7, 1007.8 and 1021.8, respectively.

Based on the above analysis, the ion with m/z 1056.9 contained in Peak I had a MW 18 Da which was higher than that of MC-RR. MC-RR has the structure of cyclo-D-alanine-arginine-D-erythro- β -methyl-

Fig. 3 HPLC profile of the biodegradation of MC-RR catalyzed by intracellular enzymes of *Sphingopyxis* sp. USTB-05.

D-isospartic acid-arginine-Adda-D-isoglutamic acid-N-methyldehydroalanine (Fig. 6). Adda is a characteristic β -amino acid (2S, 3S, 8S, 9S)-3-amino-9-methoxy-2,6,8-trimethyl-10-phenyldeca-4(E), 6(E)-dienoic acid, which is essential to the toxicity of MC-RR. Previous studies revealed that Adda-Arg was the bond that was easily attacked during the enzymatic degradation by many bacteria (Bourne et al., 1996, 2001; Imanishi et al., 2005). Ion with m/z 1056.9 may thus be derived from MC-RR by opening the ring through breaking the Adda-Arg bond and combining one hydrogen on the NH_2 group of Adda and a hydroxyl on the carboxyl group of arginine. MC-RR was turned into a linear (NH_2 -Adda-Glu-Mdha-Ala-Arg-MeAsp-Arg-OH) as intermediate product-a (Fig. 6). The processes lead a higher MW of product-a (18 Da) compare to MC-RR. Ion with m/z 905.6 had MW 133 Da lower than that of MC-RR, which coincided with the loss of the terminal phenylethylmethoxy group and the amino NH_2 group from the Adda after Adda-Arg bond breaking. This is a common loss fragment of MCs due to the radical fragmentation rearrangement (Namikoshi et al., 1992). The ion with m/z 1021.8 had a MW 36 Da which was lower than that of product-a, corresponding to two H_2O loss from product-a. Studies with ^{18}O -labeled peptides showed that dehydration can occur from the free carboxylic groups of the participant amino acids, hydroxyl substituted bonds and even the backbone of the peptides (Antoniou et al., 2008). The most basic sites and therefore easily protonated site in MC-LR are the guanidine portion of arginine, as supported by computation studies for the lowest energy conformation of the singly protonated MC-LR (Yuan et al., 1999). Thus, product-a may go through two dehydrations

arising from the combination of two hydrogen of amino group on arginine with adjacent hydroxyl on carboxy to form the linear MC-RR with the two small peptide rings as the final product-b (Fig. 6). The peaks of other two ions with m/z 1007.8 and 991.7 were not separated from that of product-b in UV chromatogram, which indicated that they may be the fragments of product-b due to the radical fragmentation rearrangement. Because the ion with m/z 1007.8 had lower MW (14 Da) than product-b, a demethylation might happen. The previous studies reported that the methoxy group of the Adda amino acid can be easily protonated, where the fragmentation is easy to happen (Yuan et al., 1999). Ion with m/z 1007.8 thus might be formed by the demethylation from methoxy group of Adda. Ion with m/z 991.7 had a lower MW (30 Da) than product-b, except one CH_3 loss, and one NH_2 loss from Adda group may be happened. The structure of product-b was gotten by conjecture based on the above results. More studies should be carried out to characterize the structure of product-b by other methods such as nuclear magnetic resonance. The possible degradation pathways of MC-RR by *Sphingopyxis* sp. USTB-05 are illustrated in Fig. 6.

2.4 General impact in natural lakes

Biodegradation is one of the essential processes for the reduction of MCs in natural eutrophic lakes and reservoirs. MCs degrading bacteria are distributed all over the world. Studies on the enzymatic pathways of various isolated strains for MCs are important to understand their fate in natural environment. Because of floating, algae toxins release is often subject to aerobic biodegradation in time scale of minutes to hours in the water column, which

Fig. 4 LC/MS profile of MC-RR and its biodegradation products. (a) total ion chromatogram (TIC); (b) UV chromatogram; (c) MS spectrum for MC-RR; (d) MS spectrum for peak I.

Fig. 5 MALDI-TOF MS profile of final biodegradation products of MC-RR catalyzed by the enzymes of *Sphingopyxis* sp. USTB-05.

may greatly change their molecular structure and toxicity. Because of the complexity of natural bacteria in the natural lakes, detail information on biodegradation pathways are often obtained through studies on pure strains. Up to now, as to the biodegradation pathways of MC-

RR, studies are focused on two strains, i.e., *Sphingomonas* sp. B-9 and CBA4. The mechanisms of Adda-Arg bond breaking for *Sphingomonas* sp. B-9 and demethylation for *Sphingomonas* sp. CBA4 were found in the enzymatic degradation processes of MC-RR. Here, our study showed that the degradation pathways of MC-RR by the strain *Sphingopyxis* sp. USTB-05 were different from the two strains mentioned above. Except for the Adda-Arg bond breaking, the mechanism of dehydration was involved in the MC-RR enzymatic degradation processes by *Sphingopyxis* sp. USTB-05 which made the structure of the final product different from that reported in the previous studies (Imanishi et al., 2005; Amé et al., 2006). The fact, MC-RR biodegradation mechanisms may vary with different bacteria implies that the same MC-RR may go through different biodegradation pathways and hence different fate in various lakes. Adda-Arg ring opening is essential for the detoxification (Bourne et al., 1996). Studies should be taken into consideration for the risk assessment of MC-RR in natural lakes. Studies on cooperative and antagonistic effects between different strains are also needed for understanding biodegradation pathways of MC-RR and their impacts on the toxicity in the aquatic environment.

Fig. 6 Proposed biodegradation pathways of MC-RR by the enzymes of *Sphingopyxis* sp. USTB-05.

3 Conclusions

(1) *Sphingopyxis* sp. USTB-05 isolated from Lake Dianchi has a strong ability in the biodegradation of MC-RR at the rate of 16.7 mg/L per day.

(2) In the enzymatic biodegradation pathways of MC-RR by *Sphingopyxis* sp. USTB-05, the Adda-Arg peptide bond of MC-RR was cleaved firstly to form linear one as the intermediate product (product-a), and then dehydrated by combining two hydrogen of amino group on arginine with adjacent hydroxyl on carboxy to form the final degradation product-b.

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